

Mass preparation of AFPI, AFPIII, and AFGP from Japanese fish

NICHIREI



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Abstract

Freezing is initiated by generation of numerous embryo ice crystals in water, which grow and merge together to form their multicrystalline state, the ordinary ice block. Antifreeze protein (AFP) is capable of binding to these ice crystals to inhibit their growth, and disturbs such general ice formation. This mechanism is expected to solve many technical problems with regard to the frozen storage of water-containing materials, such as foods and tissues. The fish-derived AFP can also bind to the lipid bilayer to prolong the lifetime of a cell, which is applicable to the short-term hypothermic cell preservation. Natural fish AFP is a mixture of 2–13 isoforms that function together far more effectively than any single isoform. We have been therefore trying to develop mass preparation method of natural fish AFP (the mixture), and now AFPI, AFPIII, and AFGP samples are available from NICHIREI CORPORATION, Japan (E-mail to s.tsuda@aist.go.jp or directly to N1000X016@nichirei.co.jp). Each AFP sample is highly purified (>95%) and contains neither cations nor buffer detergents. The samples are also sterilized using 0.22 μm syringe filter, so that directly applicable to any kind of experiment including medical tests. AFPII sample will also be released soon.

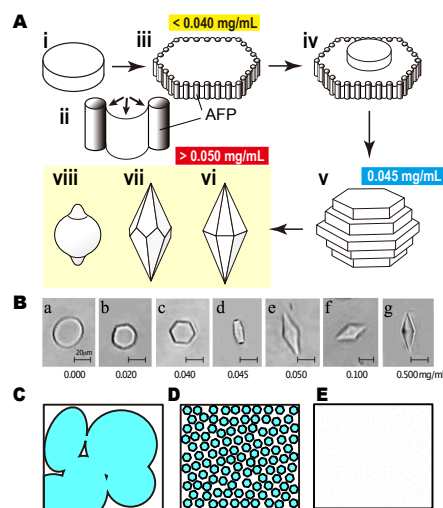
Products

We have been improved the methods to purify the AFP samples from fish muscle homogenates (Nishimiya, 2008). Our AFPI, AFPIII, and AFGP show 800, 50, and 1,600 mg/mL of high water solubility, and their maximum TH values are 3.2, 2.1, and 5.0°C, respectively. They are also recovered from heating to 95°C. In addition to these highly purified AFPs, crude products of approximately 20% purity can also be supplied on request.

AFPI	AFPIII	AFGP
Pluronectidae	Zoarcidae	Gadidae
Ala-rich, 11-res.repeats 3.3 kDa	Unbiased, SP and QAE-groups 7 kDa	 2.6–3.4 kDa
 α-Helix	 β-Sandwich	 Polyproline II helix
*AFPI	AFPIII	AFGP

Functions

A single ice crystal consists of water molecules in a hexagonal arrangement, while in solution it forms a disk-like shape (A-i). In AFP solutions, ice crystal growth is only allowed between the bound AFPs (A-ii) according to the Gibbs-Thomson effect. The resultant convex ice front sandwiched between the AFPs is terminated its growth to form a flat, AFP-accumulated surface (A-iii). This process transfers the disk-shaped ice crystal into a hexagonal plate, and allows generation of a new disk on that plate through a mechanism of 2D-nucleation. Repeated AFP binding and new disk generation causes successive stacking of smaller hexagonal ice plates in the direction of the c-axis (A-iv&v), forming a hexagonal bipyramid (A-vi) onto which millions of AFPs are adsorbed. This unique ice crystal is further modified into its derivative forms such as the hexagonal trapezohedron (A-vii) or the lemon-like shape (A-viii). The ice hexagonal plate remains unchanged and is not modified into a bipyramid when AFP concentration is below 40–50 μg/mL (depends on the condition), which we termed the “critical ice-shaping concentration” (CISC) (Mahatabuddin et al. 2016). The general disk-shape ice crystals (A-i) undergoes ice recrystallization and forms multicrystalline state by freezing (C). This ice expands the volume and physically destroys inside texture of all of the frozen materials. AFP is capable of inhibiting growth of the ice crystals to form ice slurry (D). If we can minimize the size of each ice crystal ultimately small, the frozen state will become noncrystalline glass-like state (E). That is, AFP should be able to freeze and preserve the water-containing materials by filling their inside with numerous tiny ice crystals. This can be realized with a home freezer (-20 degC) without LN₂ nor deep freezer.



Application 1

AFP exhibited a gel protection function, which made us to develop “gelation freezing method” to fabricate porous materials, for which AFP has a significant contribution (Fukushima et al. 2013). A solution of gelatin, ceramic powder, and AFP is initially cooled to form a gel structure, and placed on a freezing plate to induce unidirectional freezing. Because AFP binds to the surface of the elongating ice crystals, extremely sharpened and uniformly aligned ice needles are created in the frozen gel. After sintering it at 1,000°C, a ceramic material containing unidirectionally-aligned dendritic pores was created.

0.5% (w/v) Agarose gel after freeze-thawing

AFP = 0.01 mg/ml (0.001w/v %), 0.02 mg/ml (0.002 w/v %), 0.05 mg/ml (0.005w/v %), 0 mg/ml

Unidirectional freezing
Gelatin, Ceramics Powder, AFP
Ice needles
1,000°C

Application 2

Cell protection effect of AFP has been suggested in both cryo- and hypothermic condition. The former used very diluted solution (below CISC) so as not to create ice bipyramids. In nature, AFP functions with glucose, glycerol, lipids, ions, minerals, peptides, etc. AFP preservation will be maximized by using optimal combinations of these substances, which will differ between the applications.

Cells	Freezing (Cryo) preservation	References
Rat kidney	TmAFP -4°C	61.5 μM Tomally 2017
Mouse ovary	AFPIII, FTBP vitrification	10 mg/ml Kim 2017
Bovine oocyte	AFGP vitrification	1 mM Liang 2016
Buffalo sperm	AFGP -196°C	1,000 μg/ml Qadee 2016
Marine diatom	LeIBP -196°C	100 μg/ml Lee 2015
Buffalo sperm	AFPIII -196°C	20 mg/ml Qadee 2014
Rabbit sperm	AFPIII -196°C	1 μg/ml Nishijima 2014
Rabbit embryo	AFPIII -196°C	500 μg/ml Nishijima 2014
Seabream sperm	AFPI, III -196°C	1 μg/ml Beirao 2012
Mouse oocyte	AFPIII vitrification	500 μg/ml Jo 2012
Red blood cell	LeIBP -196°C	800 μg/ml Lee 2012
Mouse oocyte	AFPIII vitrification	500 μg/ml Jo 2011
Seabream embryo	AFPI -10°C	N/A Robles 2007
Human blood cell	syAFGP -196°C	500 μg/ml Matsumoto 2006
Rat islet	syAFGP -196°C	500 μg/ml Matsumoto 2006
Bovine sperm	AFPI -196°C	1 μg/ml Prathalingam 2006
Red blood cell	AFPI, II, III -196°C	340 μg/ml Chao 1996
Rat liver	AFGP -3°C	1 mg/ml Rubinsky 1994
Red blood cell	AFPI -196°C	150 μg/ml Carpenter 1992
Mouse embryo	AFGP vitrification	40 mg/ml Rubinsky 1992
Cells	Non-freezing preservation	References
Bovine embryo	AFPI, III +4°C	10 days 10 mg/ml Ideta 2015
Rat insulinoma cell	AFPI, III +4°C	5 days 10 mg/ml Kamijima 2013
Mouse spermatozoa	AFPIII +4°C	5 days 1 mg/ml Kiga 2011
Rat neurons	AFPI +4°C	8 hrs 10 mg/ml Rubinsky 2010
Human hepatoma cell	AFPIII +4°C	72 hrs 10 mg/ml Hirano 2008
Rat heart	AFPI, III -1.3°C	32 hrs 15 mg/ml Amir 2004
Carp Spermatozoa	AFGP +4°C	5 days 10 mg/ml Karanova 2002
Human platelet	AFPI, III +4°C	4 days 10 mg/ml Bagui 1997
Human oocyte	AFGP +4°C	21 days 1 mg/ml Tablin 1996
Rat liver	AFPI, III +4°C	48 hrs 1 mg/ml Rubinsky 1993
Bovine oocyte	AFPI, III +4°C	24 hrs 20 mg/ml Lee 1992
pig oocyte	AFGP +4°C	24 hrs 40 mg/ml Rubinsky 1991

Summary

Highly purified natural fish AFPI, III, and AFGP containing neither salts nor buffer detergents are now supplied from NICHIREI CORPORATION, Japan. Ref. Nishimiya et al. Synthesiology 2008, 1 (1) 4-17; Mahatabudding et al. Cryobiology and Cryotechnology 2016, 62 (2) 95-103; Fukushima et al. 2013, J Am Ceram Soc 96, 1029-1031. This research was supported by JSPS (15K13760).

* AFPI is “BpAFP” introduced in Mahatabuddin et al. 2017 Concentration-dependent oligomerization of an alpha-helical antifreeze polypeptide makes it hyperactive, Scientific Reports 7,42501.
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